

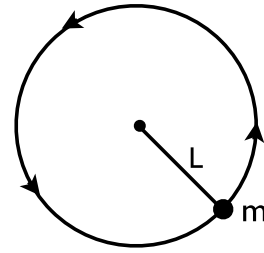
## Solutions

Physics 116 - Second In-Class Practice Exam

Profs. Schnetzer and Kloet

1. A ball of mass  $m$ , at one end of a string of length  $L$  moves in a vertical circle just fast enough to prevent the string from going slack at the top of the circle. The speed of the ball at the bottom of the circle is:

- a)  $\sqrt{2gL}$
- b)  $\sqrt{3gL}$
- c)  $\sqrt{4gL}$
- d)  $\sqrt{5gL}$
- e)  $\sqrt{7gL}$



2. Each of the four corners of a square with edge  $a$  is occupied by a point mass  $m$ . There is a fifth mass, also  $m$ , at the center of the square. The work that must be done to remove the mass from the center to a point very far away is:

- a)  $4Gm^2/a$
- b)  $-4Gm^2/a$
- c)  $4\sqrt{2}Gm^2/a$
- d)  $-4\sqrt{2}Gm^2/a$
- e)  $4Gm^2/a^2$

3. Solid  $A$  with mass  $m$  is at its melting point  $T_A$ . It is placed in thermal contact with solid  $B$  also with mass  $m$ . Solid  $B$  is initially at temperature  $T_B$  ( $T_B > T_A$ ). The combination is thermally isolated.  $A$  has latent heat of fusion  $L$  and when it is melted has specific heat capacity  $c_A$ . Solid  $B$  has specific heat capacity  $c_B$ . If  $A$  completely melts, the final temperature of both  $A$  and  $B$  is:

- a)  $(c_A T_A + c_B T_B - L)/(c_A + c_B)$
- b)  $(c_A T_A - c_B T_B + L)/(c_A + c_B)$
- c)  $(c_A T_A - c_B T_B - L)/(c_A + c_B)$
- d)  $(c_A T_A + c_B T_B + L)/(c_A + c_B)$
- e)  $(c_A T_A + c_B T_B + L)/(c_A - c_B)$

4. A  $2 \text{ m}^3$  weather balloon is filled with helium at  $P = 1 \text{ atm}$  and  $T = 27^\circ \text{ C}$ . At an elevation of 20,000 ft., the pressure is down by a factor of 2 and the temperature is down to  $T = -48^\circ \text{ C}$ . What is the volume of the balloon at this elevation?

- a)  $3 \text{ m}^3$
- b)  $4 \text{ m}^3$
- c)  $2 \text{ m}^3$
- d)  $2.5 \text{ m}^3$
- e)  $5.3 \text{ m}^3$

5. An air bubble doubles in volume as it rises from the bottom of a lake to the top. Assuming that the lake is at uniform temperature, the depth of the lake is:

- a) 21 m
- b) 0.76 m
- c) 4.9 m
- d) 10 m
- e) 0.99 m

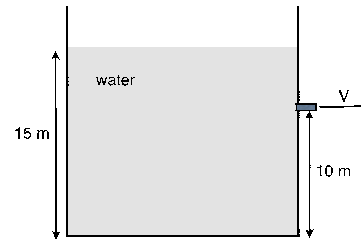
6. Two ideal monatomic gases are in thermal equilibrium with each other. Gas  $A$  is composed of molecules with mass  $m$  while gas  $B$  is composed of molecules with mass  $4m$ . The ratio of the root mean square velocity of gas  $A$  to the root mean square velocity of gas  $B$  is:

- a)  $1/4$
- b)  $1/2$
- c) 1
- d) 2
- e) 4

7. For constant-volume processes the heat capacity of gas  $A$  is greater than the heat capacity of gas  $B$ . We conclude that when they absorb the same energy as heat at constant volume:
- a) the temperature of  $A$  increases more than the temperature of  $B$
  - b) the temperature of  $B$  increases more than the temperature of  $A$
  - c) the internal energy of  $A$  increases more than the internal energy of  $B$
  - d) the internal energy of  $B$  increases more than the internal energy of  $A$
  - e)  $A$  does more positive work than  $B$
8. A balloon is released from a tall building. The total mass of the balloon including the enclosed gas is 2 kg. Its volume is  $5.0 \text{ m}^3$ . The density of air is  $1.3 \text{ kg/m}^3$ . Will the balloon rise, fall or remain stationary and why?
- a) The balloon will fall because its density is greater than that of air
  - b) The balloon will remain stationary because its density is less than that of air
  - c) The balloon will rise because the upward buoyant force is greater than the weight
  - d) The balloon will fall because the upward buoyant force is less than its weight
  - e) The question cannot be answered because the density of the gas inside the balloon is not given

9. A large tank is filled with water to a depth of 15 m. A spout located 10.0 m above the bottom of the tank is then opened as shown in the drawing. With what speed will water emerge from the spout?

- a) 3.1 m/s
- b) 9.9 m/s
- c) 14 m/s
- d) 17 m/s
- e) 31 m/s

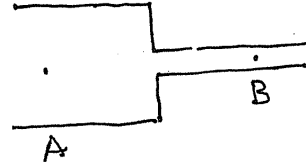


10. Alyssa went to her outside faucet to fill a  $0.032 \text{ m}^3$  bucket with water to wash her car. Water exits with a speed of  $0.64 \text{ m/s}$ . The radius of the faucet is  $0.0075 \text{ m}$ . How long does it take to fill the bucket completely? (Assume that the water is non-viscous.)

- a) 9.0 s
- b) 17.0 s
- c) 190 s
- d) 280 s
- e) 890 s

11. The density of the liquid flowing through the horizontal pipe in the drawing is  $1500 \text{ kg/m}^3$ . The speed of the fluid at point A is  $5.5 \text{ m/s}$  while at point B it is  $8.0 \text{ m/s}$ . What is the difference in pressure,  $(P_B - P_A)$ , between points B and A?

- a)  $+3.8 \times 10^3 Pa$
- b)  $-1.9 \times 10^3 Pa$
- c)  $+5.0 \times 10^4 Pa$
- d)  $-2.5 \times 10^4 Pa$
- e)  $-7.6 \times 10^5 Pa$



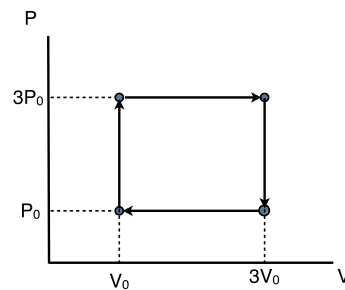
12. The pressure of an ideal gas is doubled during a process in which the energy lost as heat by the gas equals the work done on the gas. As a result, the volume is:
- a) doubled
  - b) halved
  - c) unchanged
  - d) more information is needed to answer
  - e) the process is impossible

13. A monatomic ideal gas has molar specific heat  $C_p$  at constant pressure. When the temperature of  $n$  moles of the gas is increased by  $\Delta T$  the increase in the internal energy of the gas is:

- a)  $nC_p\Delta T$
- b)  $n(C_p + R)\Delta T$
- c)  $n(C_p - R)\Delta T$
- d)  $n(2C_p + R)\Delta T$
- e)  $n(2C_p - R)\Delta T$

14. An ideal gas is taken through the cycle shown in the figure. What is the net amount of heat that flows into the gas in each cycle?

- a)  $8P_0V_0$
- b)  $-8P_0V_0$
- c)  $4P_0V_0$
- d)  $-4P_0V_0$
- e) zero



15. An inventor claims to have a heat engine that has an efficiency of 40% when it operates between a high temperature reservoir of  $150^{\circ}\text{C}$  and a low temperature reservoir of  $30^{\circ}\text{C}$ . This engine:
- a) violates more than one of the laws of thermodynamics
  - b) violates only the first law of thermodynamics
  - c) violates only the second law of thermodynamics
  - d) violates only the third law of thermodynamics
  - e) does not violate any of the laws of thermodynamics